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2. Il candidato illustri le principali esperienze di studio e/o professionali indicate nel Curriculum vitae et studiorum in formato europeo;
3. Il candidato legga e traduca il brano in inglese, contenuto nella busta ed evidenziato in giallo;
4. Il candidato illustri le principali caratteristiche del pacchetto Microsoft Office



Marine Pollution Bulletin

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Review

Microplastics as contaminants in the marine environment: A review

Matthew Cole a,*, Pennie Lindeque a, Claudia Halsband b, Tamara S. Galloway c

ARTICLE INFO

Keywords: Microplastics Marine litter Plastic debris Priority pollutant

ABSTRACT

Since the mass production of plastics began in the 1940s, microplastic contamination of the marine environment has been a growing problem. Here, a review of the literature has been conducted with the following objectives: (1) to summarise the properties, nomenclature and sources of microplastics; (2) to discuss the routes by which microplastics enter the marine environment; (3) to evaluate the methods by which microplastics are detected in the marine environment; (4) to assess spatial and temporal trends of microplastic abundance; and (5) to discuss the environmental impact of microplastics. Microplastics are both abundant and widespread within the marine environment, found in their highest concentrations along coastlines and within mid-ocean gyres. Ingestion of microplastics has been demonstrated in a range of marine organisms, a process which may facilitate the transfer of chemical additives or hydrophobic waterborne pollutants to biota. We conclude by highlighting key future research areas for scientists and policymakers.

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1. Introduction

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Whilst the societal benefits of plastic are far-reaching (Andrady and Neal, 2009), this valuable commodity has been the subject of increasing environmental concern. Primarily, the durability of plastic that makes it such an attractive material to use also makes it highly resistant to degradation, thus disposing of plastic waste is problematic (Barnes et al., 2009; Sivan, 2011). Exacerbated by the copious use of throw-away "user" plastics (e.g. packaging material), the proportion of plastic contributing to municipal waste constitutes 10% of waste generated worldwide (Barnes et al., 2009). While some plastic waste is recycled, the majority ends up in landfill where it may take centuries for such material to breakdown and decompose (Barnes et al., 2009; Moore, 2008). Of particular concern are plastics that, through indiscriminate disposal, are entering the marine environment (Gregory, 2009). Despite plastics being an internationally recognised pollutant with legislation in place aimed to curb the amount of plastic debris entering the marine environment (Gregory, 2009; Lozano and Mouat, 2009), Thompson (2006) estimates up to 10% of plastics produced end up in the oceans, where they may persist and accumulate.

The impact that large plastic debris, known as 'macroplastics', can have on the marine environment has long been the subject of environmental research. The presence of macroplastics in the marine environment presents an aesthetic issue, with economic repercussions for the tourist industry, a hazard for numerous marine-industries (e.g. shipping, fishing, energy production, aquaculture) as plastic may result in entanglement and damage of equipment, and significant environmental concerns (Barnes et al., 2009; Derraik, 2002; Sivan, 2011). The environmental impact of macroplastics include: the injury and death of marine birds, mammals, fish and reptiles resulting from plastic entanglement and ingestion (Derraik, 2002; Gregory, 2009; Lozano and Mouat, 2009), the transport of non-native marine species (e.g. bryozoans) to new habitats on floating plastic debris (Barnes, 2002; Derraik, 2002; Winston, 1982), and the smothering of the seabed, preventing gas-exchange and creating artificial hard-grounds, resulting from sinking plastic debris (Gregory, 2009; Moore, 2008).

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2. Microplastics

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Plastics that are manufactured to be of a microscopic size are defined as primary microplastics. These plastics are typically used in facial-cleansers and cosmetics (Zitko and Hanlon, 1991), or as air-blasting media (Gregory, 1996), whilst their use in medicine as vectors for drugs is increasingly reported (Patel et al., 2009). Under the broader size definitions of a microplastic, virgin plastic production pellets (typically 2–5 mm in diameter) can also be considered as primary microplastics, although their inclusion within this category has been criticised (Andrady, 2011; Costa et al., 2010).

Microplastic "scrubbers", used in exfoliating hand cleansers and facial scrubs, have replaced traditionally used natural ingredients, including ground almonds, oatmeal and pumice (Derraik, 2002; Fendall and Sewell, 2009). Since the patenting of microplastic scrubbers within cosmetics in the 1980s, the use of exfoliating cleansers containing plastics has risen dramatically (Fendall and Sewell, 2009; Zitko and Hanlon, 1991). Typically marketed as "micro-beads" or "micro-exfoliates", these plastics can vary in shape, size and composition depending upon the product (Fendall and Sewell, 2009). For example, Gregory (1996) reported the presence of polyethylene and polypropylene granules (<5 mm) and polystyrene spheres (<2 mm) in one cosmetic product. More recently, Fendall and Sewell (2009) reported an abundance of irregularly shaped microplastics, typically <0.5 mm in diameter with a mode size <0.1 mm, in another cosmetic product.

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Secondary microplastics describe tiny plastic fragments derived from the breakdown of larger plastic debris, both at sea and on land (Ryan et al., 2009; Thompson et al., 2004). Over time a culmination of physical, biological and chemical processes can reduce the structural integrity of plastic debris, resulting in fragmentation (Browne et al., 2007).

Over prolonged periods, exposure to sunlight can result in photo-degradation of plastics; ultraviolet (UV) radiation in sunlight causes oxidation of the polymer matrix, leading to bond cleavage (Andrady, 2011; Barnes et al., 2009; Browne et al., 2007; Moore, 2008; Rios et al., 2007). Such degradation may result in additives, designed to enhance durability and corrosion resistance, leaching out of the plastics (Talsness et al., 2009). The cold,

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The objectives of this review are: (1) to summarise the properties, nomenclature and sources of microplastics; (2) to discuss the routes by which microplastics enter the marine environment; (3) to evaluate the methods by which microplastics are detected in the marine environment; (4) to ascertain spatial and temporal trends of microplastic abundance; and (5) to determine the environmental impact of microplastics.

2. Microplastics

Whilst macroplastic debris has been the focus of environmental concern for some time, it is only since the turn of the century that tiny plastic fragments, fibres and granules, collectively termed "microplastics", have been considered as a pollutant in their own right (Ryan et al., 2009; Thompson et al., 2004). Microplastics have been attributed with numerous size-ranges, varying from study to study, with diameters of <10 mm (Graham and Thompson, 2009), <5 mm (Barnes et al., 2009; Betts, 2008), 2–6 mm (Derraik, 2002), <2 mm (Ryan et al., 2009) and <1 mm (Browne et al., 2007; Browne et al., 2010; Claessens et al., 2011). This inconsistency is particularly problematic when comparing data referring to microplastics, making it increasingly important to create a scientific standard (Claessens et al., 2011; Costa et al., 2010). Recently, Andrady (2011) has suggested adding the term "mesoplastics" to scientific nomenclature, to differentiate between small plastics visible to the human eye, and those only discernible with use of microscopy.

2.1. Primary microplastics

Plastics that are manufactured to be of a microscopic size are defined as primary microplastics. These plastics are typically used in facial-cleansers and cosmetics (Zitko and Hanlon, 1991), or as air-blasting media (Gregory, 1996), whilst their use in medicine as vectors for drugs is increasingly reported (Patel et al., 2009). Under the broader size definitions of a microplastic, virgin plastic production pellets (typically 2–5 mm in diameter) can also be considered as primary microplastics, although their inclusion within this category has been criticised (Andrady, 2011; Costa et al., 2010).

Microplastic "scrubbers", used in exfoliating hand cleansers and facial scrubs, have replaced traditionally used natural ingredients, including ground almonds, oatmeal and pumice (Derraik, 2002; Fendall and Sewell, 2009). Since the patenting of microplastic scrubbers within cosmetics in the 1980s, the use of exfoliating cleansers containing plastics has risen dramatically (Fendall and Sewell, 2009; Zitko and Hanlon, 1991). Typically marketed as "micro-beads" or "micro-exfoliates", these plastics can vary in shape, size and composition depending upon the product (Fendall and Sewell, 2009). For example, Gregory (1996) reported the presence of polyethylene and polypropylene granules (<5 mm) and polystyrene spheres (<2 mm) in one cosmetic product. More recently, Fendall and Sewell (2009) reported an abundance of irregularly shaped microplastics, typically <0.5 mm in diameter with a mode size <0.1 mm, in another cosmetic product.

Primary microplastics have also been produced for use in airblasting technology (Derraik, 2002; Gregory, 1996). This process involves blasting acrylic, melamine or polyester microplastic scrubbers at machinery, engines and boat hulls to remove rust and paint (Browne et al., 2007; Derraik, 2002; Gregory, 1996). As these scrubbers are used repeatedly until they diminish in size and their cutting power is lost, they will often become contaminated with heavy metals (e.g. Cadmium, Chromium, Lead) (Derraik, 2002; Gregory, 1996).

2.2. Secondary microplastics

Secondary microplastics describe tiny plastic fragments derived from the breakdown of larger plastic debris, both at sea and on land (Ryan et al., 2009; Thompson et al., 2004). Over time a culmination of physical, biological and chemical processes can reduce the structural integrity of plastic debris, resulting in fragmentation (Browne et al., 2007).

Over prolonged periods, exposure to sunlight can result in photo-degradation of plastics; ultraviolet (UV) radiation in sunlight causes oxidation of the polymer matrix, leading to bond cleavage (Andrady, 2011; Barnes et al., 2009; Browne et al., 2007; Moore, 2008; Rios et al., 2007). Such degradation may result in additives, designed to enhance durability and corrosion resistance, leaching out of the plastics (Talsness et al., 2009). The cold,